

Technical Bulletin**TERWET[®] 5522 Surfactant****PRODUCT DESCRIPTION**

TERWET[®] 5522 surfactant a water-soluble, nonionic surface active agent which is compatible with other nonionic surfactants and with most anionic and cationic surfactants. The product is a white, waxy solid at room temperature.

APPLICATIONS

- Emulsifier
- Solubilizer
- Animal Health/Teat Dip

SALES SPECIFICATIONS

<u>Property</u>	<u>Specifications</u>	<u>Test Method*</u>
Appearance, 150°F	Clear to slightly turbid liquid, substantially free of foreign matter	ST-30.1
Acid Value, mg KOH/g	0.2 max.	ST-31.41, F
Color, Pt-Co, 150°F	150 max.	ST-30.12
Hydroxyl Number, mg KOH/g	46 - 52	ST-31.39
Water, wt%	0.3 max.	ST-31.53

*Methods of Test are available from Huntsman Corporation upon request.

ADDITIONAL INFORMATION**Regulatory Information**

40 CFR180 910, 930, and 940a
See SDS for all regulatory information.

Shelf Life

The product should retain its conformance to sales specifications for a period of at least two years from date of manufacture if the product is stored at less than 100°F in its undamaged, unopened, factory packaged container.

In general, the user should determine the suitability of any chemical compound, no matter what the shelf life or length of time of storage. Each user should conduct a sufficient investigation to establish the suitability of any product for his intended use.

Typical Properties

Flash Point, PMCC, °F	487
Flash Point, PMCC, °C	253
Melting point, °F	103
Melting point, °C	39
Density, g/ml, 60°C (140°F)	1.0309
Weight, lbs/US gal, 60°C (140°F)	8.59
Viscosity, kinematic cSt, 60°C (140°F)	46
HLB Value	16.6

TOXICITY AND SAFETY

For information on the toxicity and safe handling of this product, read the Safety Data Sheet prior to use of the product.

HANDLING AND STORAGE

TERWET[®] 5522 surfactant may be satisfactorily stored in carbon steel tanks using steel pipes and pumps. Caution must be exercised, however, to keep the material in the anhydrous state to prevent severe corrosion to the carbon steel tank and related equipment. A drier on the breathing nozzle is recommended to help maintain anhydrous conditions in the storage tank.

For longer term color stability, it is recommended that the product be stored under an inert atmosphere.

Low pressure steam coils in storage tanks and steam tracing of transfer lines should be provided in cases where low environmental temperatures may make pumping of the product difficult.

SHIPPING DATA

Product is available in tank cars and tank trucks. Small evaluation samples can be obtained by contacting any Huntsman Performance Products sales office.

BIODEGRADABILITY AND ENVIRONMENTAL SAFETY

Linear alcohol ethoxylates undergo rapid and extensive biodegradation under both laboratory and environmental conditions. Their mineralization to CO₂ and water (ultimate biodegradation) is essentially complete during biological wastewater treatments at warm to cold water temperatures. They are degraded by bacteria in rivers, lakes, groundwater and sediment as well.

The major mechanism of biodegradation is cleavage of the ethoxylate chain from the alkyl group with oxidation of the latter to fatty acid. The fatty acid degrades more rapidly than the ethoxylate chain, which is broken down by sequential oxidation and removal of ethoxylate units.

Alcohol ethoxylates begin to lose their toxicity toward aquatic organisms as soon as biodegradation begins. Water containing degraded surfactant has been shown not to adversely affect fish, invertebrates and algae. Thus, while alcohol ethoxylates are toxic to aquatic organisms, in the event of a spill into a waterway any acute effects would be limited in area and time.

Linear alcohol ethoxylates pose no serious threat to the environment. They do not accumulate in any environmental compartment and are found, if at all, only at concentrations below chronic effects levels.

General References

1. Swisher, R. D., Surfactant Biodegradation, Marcel Dekker, 1987.
2. Talmage, S. S., Environmental and Human Safety of Major Surfactants: Alcohol Ethoxylates and Alkylphenol Ethoxylates, a report to the Soap and Detergent Association, Lewis Publishers, 1994.

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